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Reaction of iron
nails with copper
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The genius of
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periodic table - Lou
Serico How To See
Germs Spread
Experiment
(Coronavirus)

Rusting - Iron +
water + oxygen =

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iron oxide Nail and
Copper II chloride
experiment

Chemistry Revision

A Iron \u0026amp;

Copper Sulphate
solution The Real
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experiment!

conditions

necessary for

rusting of iron

experiment Make

Copper Chloride (3

ways) Zinc +

Copper Sulfate

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AND CHEMICAL
CHANGES - 9
EXPERIMENT OF
RUSTING.avi

Corrosion Rate in
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CORROSION

Refraction Through
glass slab : Lateral
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conditions - YDP

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Between Iron and
Copper (II)
Chloride

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Science Sprint |
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CBSE Class 8 |
Vedantu Copper

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sulphate
crystallization |
Crystallisation |
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– Nail Lab. Purpose.
The purpose of this
investigation is to
qualitatively and
quantitatively
describe the
chemical reaction
between copper(II)

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chloride and iron.

Procedure. Day 1
(Tues or Wed) 1.

Label, then mass a
plastic cup provided
by the instructor. 2.

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Lab Purpose After
you have completed
the lab, think about
what was the

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overall point for performing the lab.
Procedure Day 1 1.
Mass a clean & labeled ___ mL beaker. 2. Put between 6.0 - 8.5 g of copper (II) chloride in the beaker.

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~~Purpose ...~~

Unit 7n: Nail Lab
Report Objective:

The objective of this lab was to observe the amount copper produced and iron used in a reaction and then produce the ratio of the copper atoms to the iron atoms.

Materials: 1.

Electronic balance

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2. 250 ml beaker 3.

100 ml beaker 4.

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~~Reflection~~

~~Assignment - Unit~~

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Nail

Lab € Apparatus and

Materials: Refer to

the reaction of iron

nails with a copper

solution assignment

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in Module 3,
Section assignment
3.4 Part F of the
Chemistry 11
course. I was
unable to do this
experiment because
I do not have
access to an
appropriate facility
that is designed for
chemistry

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Nail Experiment

Calculations Lab

Report (personal

data table,

calculations 3-steps

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questions 5 -7 in
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in Homedirs ~ HW:

a) Practice

Problems: p 327 -

3a & b, p328 - 5a &

b, and p329 - 12a -

d b) Rearranging

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